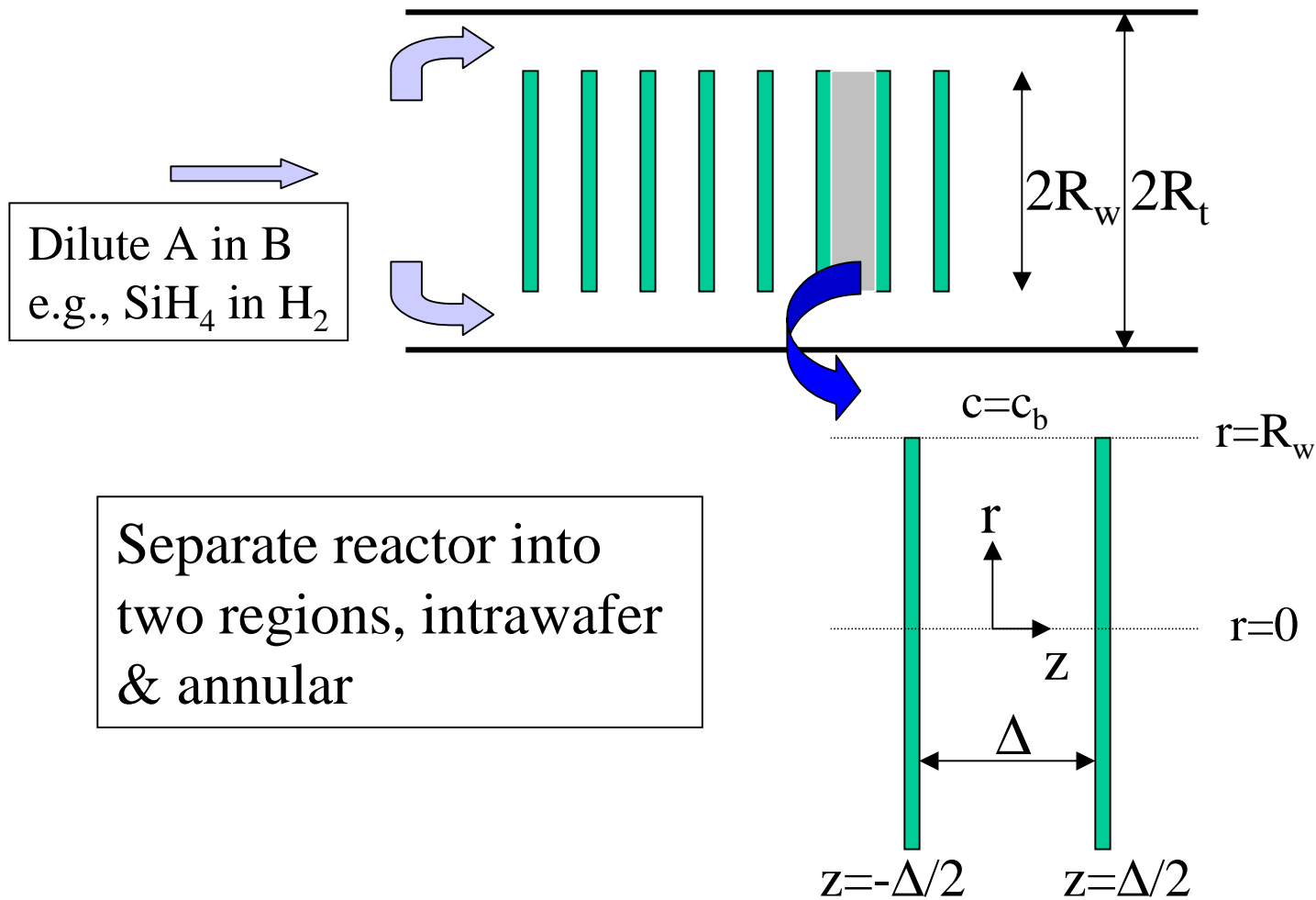


Coupled Mass Transport and Reaction in LPCVD Reactors



Continuity Eqn: Convection-Diffusion-Reaction Eqns

Assumptions

- Dilute species i in major carrier gas (e.g., H_2) $i=SiH_4$
- Isothermal
- Constant D_i and density



No rxn in the gas

$$D_i \nabla^2 c_i - \vec{u} \cdot \nabla c_i + R_i = \frac{\partial c_i}{\partial t}$$

~ 0

Assume stagnant in
the intra-wafer
region

Steady state

Equations and Boundary Conditions for the intrawafer region

$$\frac{1}{r} \frac{\partial}{\partial r} r \frac{\partial c}{\partial r} + \frac{\partial^2 c}{\partial z^2} = 0$$

$$\begin{aligned} \frac{\partial c}{\partial r} &= 0 \quad @ \quad r = 0 \\ c &= c_b \quad @ \quad r = R_w \\ D \frac{\partial c}{\partial z} &= r_s \quad @ \quad z = -\Delta/2 \\ D \frac{\partial c}{\partial z} &= -r_s \quad @ \quad z = \Delta/2 \end{aligned}$$

r_s = surface reaction rate
(net loss rate of SiH_4);
related to the deposition
rate

- If r_s is linear in c then there is an analytic solution
- If r_s is nonlinear we have to seek numerical solution

Approximate solution – “Fin Approximation”

- Take $r_s = kc$ & average over z direction
- Averaging over the small dimension (z) is called the “fin approximation”: an approximation which is very good for 2-D regions with high aspect ratio (R/Δ)

$$\int_{-\Delta/2}^{\Delta/2} \frac{1}{r} \frac{\partial}{\partial r} r \frac{\partial c}{\partial r} dz + \int_{-\Delta/2}^{\Delta/2} \frac{\partial^2 c}{\partial z^2} dz = 0$$

$$\Delta \frac{1}{r} \frac{\partial}{\partial r} r \frac{\partial C}{\partial r} + \frac{\partial C}{\partial z} \Big|_{z=\Delta/2} - \frac{\partial C}{\partial z} \Big|_{z=-\Delta/2} = 0$$

$$\Delta \frac{1}{r} \frac{\partial}{\partial r} r \frac{\partial C}{\partial r} + \left(-\frac{kC}{D} - \frac{kC}{D} \right) = 0$$

$$D \frac{1}{r} \frac{\partial}{\partial r} r \frac{\partial C}{\partial r} - \frac{2}{\Delta} kC = 0$$

$$\text{let } C(r) = \frac{\int_{-\Delta/2}^{\Delta/2} c(r, z) dz}{\int_{-\Delta/2}^{\Delta/2} dz} = \frac{1}{\Delta} \int_{-\Delta/2}^{\Delta/2} c(r, z) dz$$

$$\frac{2}{\Delta} = \frac{2\pi R_w^2}{\Delta \pi R_w^2} = \frac{\text{surface area}}{\text{volume}}$$

note

Nondimensionalization & Solution

$$\xi = \frac{r}{R_w} \quad \theta = \frac{C}{c_b}$$

$$\frac{1}{\xi} \frac{\partial}{\partial \xi} \xi \frac{\partial \theta}{\partial \xi} = \frac{2kR_w^2}{D\Delta}$$

$$\Phi^2 = \frac{2kR_w^2}{D\Delta}$$

$$\frac{\partial^2 \theta}{\partial \xi^2} + \frac{1}{\xi} \frac{\partial \theta}{\partial \xi} - \Phi^2 \theta = 0$$

Bessel's equation (Mickley Sherwood & Reid, Applied Math in ChE p. 174)

Φ is the Thiele Modulus

$$= \frac{\text{reaction rate}}{\text{diffusion rate}} = \frac{\text{diffusion time scale}}{\text{reaction time scale}}$$

$$\theta = AI_0(\Phi\xi) + BK_0(\Phi\xi)$$

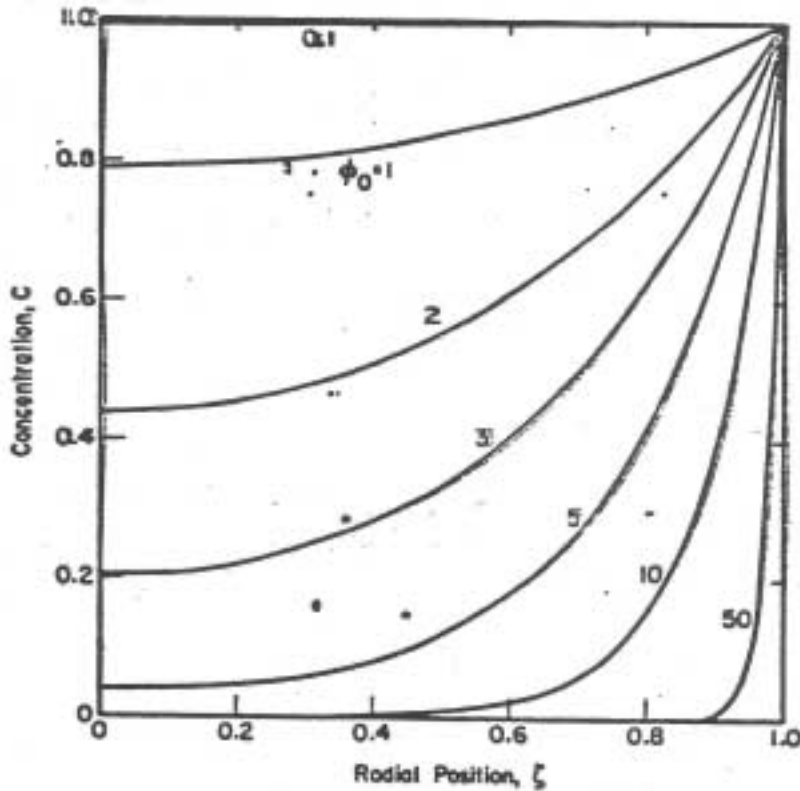
Zero order Bessel functions of the 2nd kind

Concentration profile in the intrawafer region

K_0 has logarithmic singularity at $\xi=0 \Rightarrow B = 0$

$$\theta = \frac{I_0(\Phi\xi)}{I_0(\Phi)}$$

Using $\theta=1$ at $\xi=1 \Rightarrow 1 = AI_0(\Phi)$

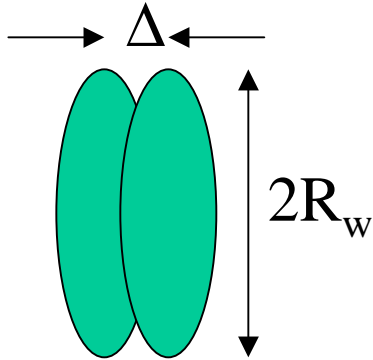


- ▶ as $\Phi \rightarrow 0$ uniformity gets better and C increases; reaction limited; diffusion faster than rxn
- ▶ as $\Phi \gg 1$ uniformity degrades and C decreases; diffusion limited; rxn faster than diffusion

$$\Phi = \frac{\text{reaction rate}}{\text{diffusion rate}}$$

Quantifying Uniformity

$$\eta = \text{“Uniformity Index”} = \frac{\text{actual deposition rate}}{\text{deposition rate if } c = c_b \text{ throughout}}$$



$$\eta = \frac{2\pi R_w \Delta \left. D \frac{dC}{dr} \right|_{r=R_w}}{2k_s c_b \pi R_w^2}$$

$$\left. \frac{dC}{dr} \right|_{r=R_w} = \frac{C_b}{R_w} \left. \frac{d\theta}{d\xi} \right|_{\xi=1} = \frac{C_b \Phi}{R_w} \frac{I_1(\Phi \xi)}{I_0(\Phi)} \Big|_{\xi=1} = \frac{C_b \Phi}{R_w} \frac{I_1(\Phi)}{I_0(\Phi)}$$

$$\eta = \frac{2\pi R_w \Delta D c_b \Phi I_1(\Phi)}{2k_s c_b \pi R_w^3 I_0(\Phi)}$$

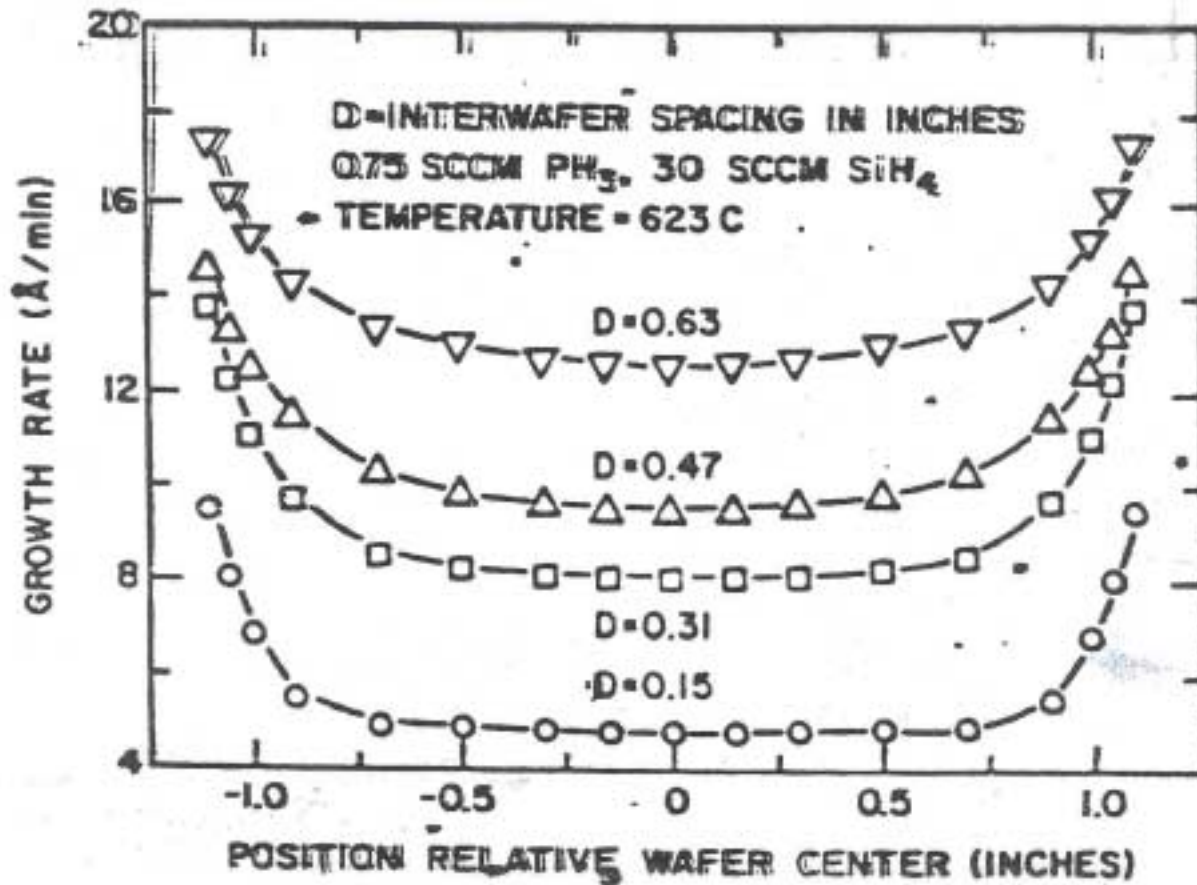


$$\eta = \frac{2 I_1(\Phi)}{\Phi I_0(\Phi)} = f(\Phi)$$

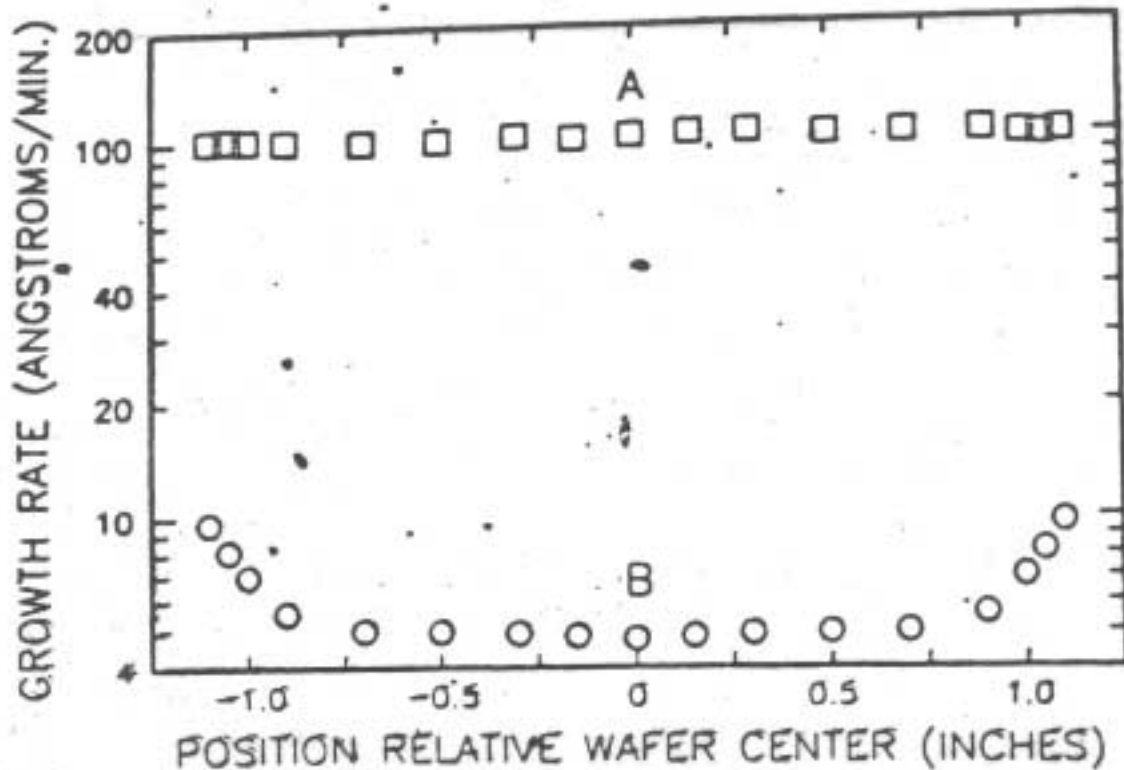
$\Phi \ll 1 \Rightarrow \eta \rightarrow 1 \Rightarrow$ uniform deposition (rxn limited)

$\Phi \gg 1 \Rightarrow \eta \rightarrow 0 \Rightarrow$ nonuniform deposition (diffusion limited)

Growth Rate distribution as a function of wafer spacing



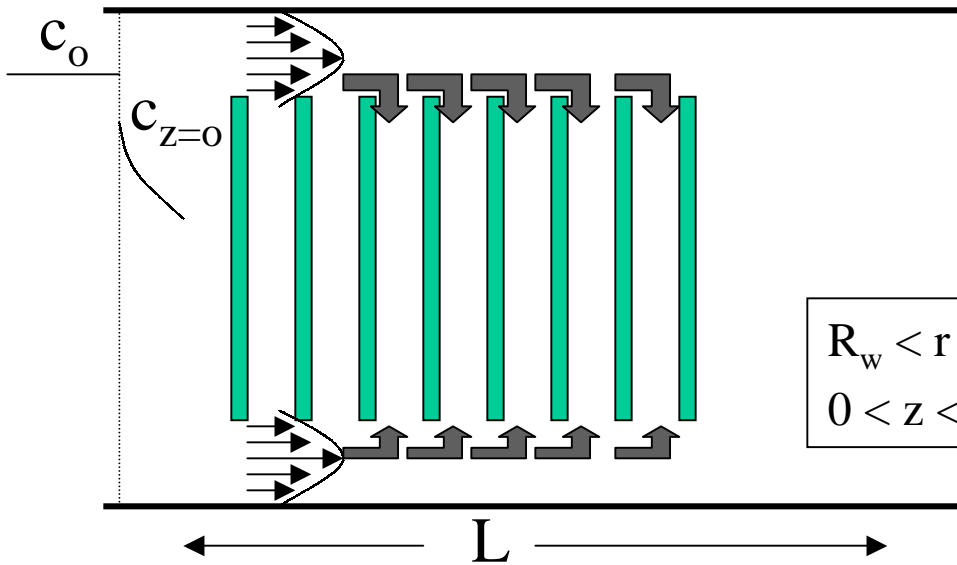
Growth Rate distribution as a function of Φ



A $\Phi \ll 1 \Rightarrow \eta \rightarrow 1 \Rightarrow$ uniform deposition (rxn limited)

B $\Phi \gg 1 \Rightarrow \eta \rightarrow 0 \Rightarrow$ nonuniform deposition (diffusion limited)

Annular Region



$$D_i \nabla^2 c_i - \vec{u} \cdot \nabla c_i + R_i = \frac{\partial c_i}{\partial t}$$

$$\begin{aligned} R_w < r < R_t \\ 0 < z < L \end{aligned}$$

No homogeneous
Gas phase rxns

Steady
state

Boundary Conditions:

Dankwertz Boundary
Conditions:

$$\begin{aligned} -D \frac{\partial c}{\partial r} &= r_s \quad @ \quad r = R_t \\ D \frac{\partial c}{\partial r} &= \eta k_s c \frac{2\pi R_w^2}{2\pi R_w \Delta} = \eta k_s c \frac{R_w}{\Delta} \quad @ \quad r = R_w \\ -D \frac{\partial c}{\partial z} \Big|_{z=0} &= u_z (c_o - c|_{z=0}) \quad @ \quad z = 0 \\ \frac{\partial c}{\partial z} \Big|_{z=L} &= 0 \quad @ \quad z = L \end{aligned}$$

r-averaged Eqns. in the Annular Region

$$D \left[\frac{1}{r} \frac{\partial}{\partial r} r \frac{\partial c}{\partial r} + \frac{\partial^2 c}{\partial z^2} \right] - u_r \frac{\partial c}{\partial r} - u_z \frac{\partial c}{\partial z} = 0$$

↙
0

- We are mostly interested in wafer-to-wafer changes (z variation) and do not care about profiles in the annulus.
- We will average over the r dimension and obtain a single r-averaged equation for C(z) where

$$\text{let } C(z) = \frac{\int_{R_w}^{R_t} \int_0^{2\pi} rc(r, z) dr d\theta}{\int_{R_w}^{R_t} \int_0^{2\pi} r dr d\theta} = \frac{2}{(R_t^2 - R_w^2)} \int_{R_w}^{R_t} rc(r, z) dr$$

$$\bar{u}_z = \frac{\int_{R_w}^{R_t} \int_0^{2\pi} ru_z(r) dr d\theta}{\int_{R_w}^{R_t} \int_0^{2\pi} r dr d\theta} = \frac{2}{(R_t^2 - R_w^2)} \int_{R_w}^{R_t} ru_z dr$$

Look how boundary conditions in r end up in the differential equation for $C(z)$ as if they are in the gas phase (i.e, the domain of diff eq.); surface reactions at radial walls appear as if they are gas phase reactions

$$D \int_{R_w}^{R_t} \frac{\partial}{\partial r} r \frac{\partial c}{\partial r} dr + \int_{R_w}^{R_t} \frac{\partial^2 c}{\partial z^2} r dr - \int_{R_w}^{R_t} r u_z \frac{\partial c}{\partial z} dr = 0$$

$$D \left[r \frac{\partial c}{\partial r} \right]_{r=R_w}^{r=R_t} + \frac{(R_t^2 - R_w^2)}{2} D \frac{d^2 C}{dz^2} - \frac{(R_t^2 - R_w^2)}{2} \bar{u}_z \frac{dC}{dz} = 0$$

$$- \left(R_t k_s C + \eta \frac{k_s C R_w^2}{\Delta} \right) + \frac{(R_t^2 - R_w^2)}{2} D \frac{d^2 C}{dz^2} - \frac{(R_t^2 - R_w^2)}{2} \bar{u}_z \frac{dC}{dz} = 0$$

$$D \frac{d^2 C}{dz^2} - \bar{u}_z \frac{dC}{dz} - \left[\frac{2k_s R_t}{(R_t^2 - R_w^2)} + 2\eta \frac{k_s R_w^2}{\Delta (R_t^2 - R_w^2)} \right] C = 0$$

Looks like

$$D \frac{d^2 c}{dz^2} - u_z \frac{dc}{dz} + R_{gas} = 0$$

Dimensional analysis

$$\Theta = \frac{C}{C_b} \quad \zeta = \frac{z}{L}$$

$$\frac{D}{\bar{u}_z L} \frac{d^2 \Theta}{d\zeta^2} - \frac{d\Theta}{d\zeta} - \left[\frac{2k_s R_t L}{\bar{u}_z (R_t^2 - R_w^2)} + \eta \frac{2k_s R_w^2 L}{\Delta \bar{u}_z (R_t^2 - R_w^2)} \right] \Theta = 0$$

$$Pe = \frac{\bar{u}_z L}{D} = \frac{\text{convective transport}}{\text{diffusive transport}}$$

$$Da_1 = \frac{2k_s R_t L}{\bar{u}_z (R_t^2 - R_w^2)} = \frac{\text{reaction rate (deposition on walls)}}{\text{convection rate}}$$

$$Da_2 = \frac{2k_s R_w^2 L}{\Delta \bar{u}_z (R_t^2 - R_w^2)} = \frac{\text{reaction rate (deposition on wafers)}}{\text{convection rate}}$$

$$\frac{1}{Pe} \frac{d^2 \Theta}{d\zeta^2} - \frac{d\Theta}{d\zeta} - [Da_1 + \eta Da_2] \Theta = 0$$

Boundary conditions and Solution

$$\frac{d\Theta}{d\zeta} = 0 \quad @ \quad \zeta = 1$$
$$-\frac{1}{Pe} \frac{d\Theta}{d\zeta} = (1 - \Theta) \quad @ \quad \zeta = 0$$

$$\Theta = \frac{2[(1 + \beta) \exp\{M(1 + \beta) + M(1 - \beta)\zeta\} - (1 - \beta) \exp\{M(1 + \beta)\zeta + M(1 - \beta)\}]}{(1 + \beta)^2 \exp\{M(1 + \beta)\} - (1 - \beta)^2 \exp\{M(1 - \beta)\}}$$

$$\text{where } \beta = \sqrt{1 + \frac{4(Da_1 + \eta Da_2)}{Pe}} \quad \text{and} \quad M = \frac{Pe}{2}$$

Dankwertz “continuous flow systems” *Chemical Engineering Science* **2** (1), 1 (1953).

Limiting Cases: PFR limit

➤ $Pe \rightarrow \infty$ (i.e., $D \rightarrow 0$ or $u_z \rightarrow \infty$)

small diffusion rate compared to convection

⇒ Plug flow reactor (PFR) limit

$$\beta = \sqrt{1 + \frac{4(Da_1 + \eta Da_2)}{Pe}} \rightarrow 1 \quad \text{and} \quad M \rightarrow \infty \quad Da^* = (Da_1 + \eta Da_2)$$

$$M(1 - \beta) = \frac{1}{2} Pe \left[1 - \sqrt{1 + \frac{4(Da_1 + \eta Da_2)}{Pe}} \right] = -\frac{2 Da^*}{\sqrt{1 + \frac{4 Da^*}{Pe}}} = -Da^*$$

$$\lim_{Pe \rightarrow \infty} \Theta = e^{-Da^* \zeta}$$

Limiting Cases: PFR limit

- ▶ Another way of seeing this is to look at $Pe \rightarrow \infty$ of the differential equation and boundary condition at $\zeta=0$

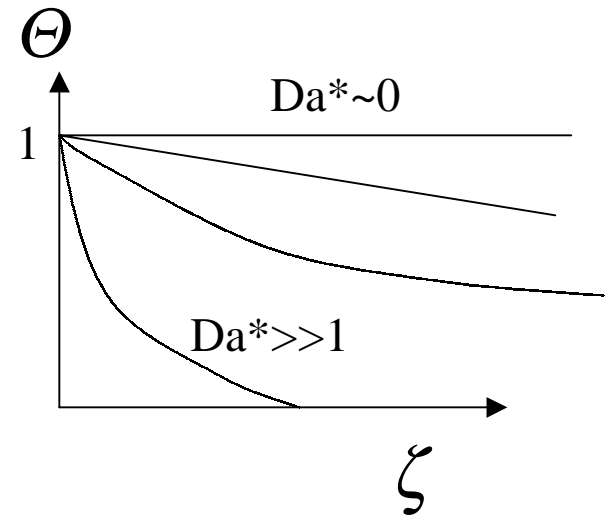
$$\frac{1}{Pe} \frac{d^2\Theta}{d\zeta^2} - \frac{d\Theta}{d\zeta} - [Da_1 + \eta Da_2]\Theta = 0$$

$$-\frac{1}{Pe} \frac{d\Theta}{d\zeta} = (1-\Theta) \quad @ \quad \zeta = 0$$

becomes $\Theta = 1 \quad @ \quad \zeta = 0$

$$\frac{d\Theta}{d\zeta} = -Da^* \Theta$$

$$\Theta = e^{-Da^*\zeta}$$



Limiting Cases: CSTR limit

➤ $Pe \rightarrow 0$ (i.e., $D \rightarrow \infty$ or $u_z \rightarrow 0$)

small flow rate (convection) compared to diffusion

⇒ Continuous stirred tank reactor (CSTR) limit

$$\frac{d^2\Theta}{d\zeta^2} - Pe \left\{ \frac{d\Theta}{d\zeta} - Da^* \Theta \right\} = 0$$

as $Pe \rightarrow 0$ $\frac{d^2\Theta}{d\zeta^2} = 0$

$$\frac{d\Theta}{d\zeta} = 0 \quad @ \quad \zeta = 1$$
$$-\frac{1}{Pe} \frac{d\Theta}{d\zeta} = (1 - \Theta) \quad @ \quad \zeta = 0$$

as $Pe \rightarrow 0$ $\frac{d\Theta}{d\zeta} = 0 \quad @ \quad \zeta = 0$

$$\lim_{Pe \rightarrow 0} \Theta = \text{constant}$$

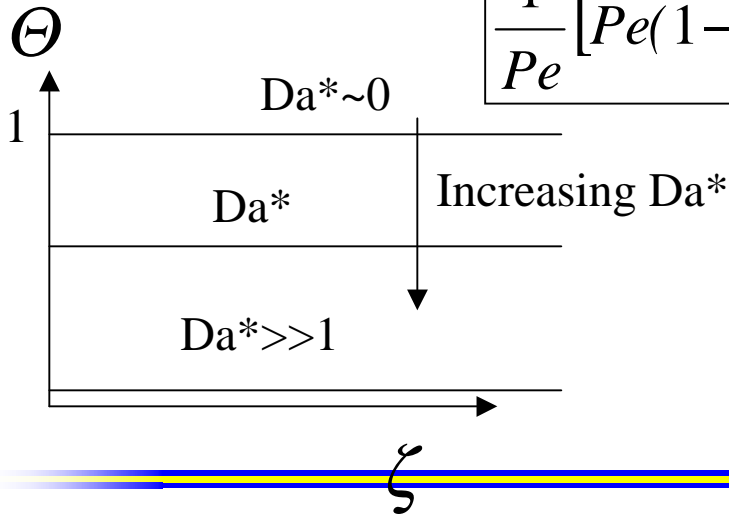
Limiting Cases: CSTR limit

- To find the concentration we average over z-direction too

$$\frac{1}{Pe} \int_0^1 \frac{d^2\Theta}{d\zeta^2} d\zeta - \int_0^1 \frac{d\Theta}{d\zeta} d\zeta - Da^* \int_0^1 \Theta d\zeta = 0$$

$$\frac{1}{Pe} \left[\frac{d\Theta}{d\zeta} \Big|_{\zeta=1} - \frac{d\Theta}{d\zeta} \Big|_{\zeta=0} \right] + \Theta(1) - \Theta(0) - Da^* \Theta = 0$$

$$\frac{1}{Pe} [Pe(1 - \Theta) - 0] - Da^* \Theta = 0$$



$$\Theta = \frac{1}{1 + Da^*}$$

← CSTR limit

Wafer-to-wafer uniformity

Uniformity index often defined as

$$U = \frac{R_{max} - R_{min}}{R_{avg}} = \frac{C_{max} - C_{min}}{C_{avg}} = \frac{\Theta_{max} - \Theta_{min}}{\Theta_{avg}}$$

- $U=0$: good uniformity
- $U \times 100 = \% \text{ is variance with respect to average}$
- As U increases uniformity degrades
- For fixed Da^* as $Pe \uparrow$ U also \uparrow
- $Pe \uparrow$ means $D \downarrow$
- Since $D \sim 1/P \Rightarrow$ lower $P \Rightarrow$ better uniformity

$$Pe = \frac{\bar{u}_z L}{D} = \frac{\text{convective transport}}{\text{diffusive transport}}$$

Deposition Rate

- Often we are interested in deposition rate, R_D in thickness/time, e.g., Å/s, Å/min, nm/s, nm/min, μm/min, etc.

deposition flux (# / cm²s) × film area (cm²) = film density (# / cm³) × \dot{V} (cm³ / s)

$$r_s A = N_f A R_D = N_f A \frac{dh}{dt}$$

$$R_D = \frac{r_s}{N_f} = \frac{kc}{N_f}$$

$$N_f = \frac{\rho_f}{M_{wf}} N_{Avogadro}$$